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### Development of a Novel Hexavalent-Chromium-Free Aluminum Sacrificial Paint

A Paper<sup>\*</sup> based on a Presentation given at NASF SUR/FIN 2022 (Rosemont, Illinois)

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*Editor's Note*: The following is a paper based on a presentation given at NASF SUR/FIN 2022, in Rosemont, Illinois on June 7, 2022, in Session 1, Aerospace / Defense Response to REACh. This is the first of two papers on the development of hexavalent-chromium-free aluminum sacrificial paints, this one from the supplier point-of-view. A concurrent second paper by Wiley, available at <a href="http://short.pfonline.com/NASF23Feb2">http://short.pfonline.com/NASF23Feb2</a>, addresses the user point-of-view in terms of the process validation work by Rolls Royce Corporation.

#### ABSTRACT

Hexavalent chromium is a known carcinogen, repro-toxin, and mutagen. Its elimination is of high importance to the aerospace industry, which has struggled to find high performing alternatives. Legacy aluminum sacrificial paints have traditionally utilized hexavalent chromium to prevent corrosion and coatings which are equal to or better than have been difficult. Many attempts have resulted in failure. In this paper, a body of results will be shared substantiating efficacy of a leading replacement. Performance testing includes cyclic-synthetic-seawater (salt spray/humidity/ambient) corrosion testing, continuous neutral salt spray corrosion testing, cyclic salt spray / humidity corrosion testing, cyclic salt spray / heat / humidity corrosion testing, heat resistance testing, fluid resistances testing, adhesion testing, conductivity/resistivity, metallography, quality, stripping, and touch up. Normalized data will show the success of a hexavalent chromium-free aluminum sacrificial paint.

#### Introduction

Hexavalent chromium is a known carcinogen, repro-toxin, and mutagen. Its elimination is of high importance to the aerospace industry which has struggled to find high performing alternatives. Legacy aluminum sacrificial paints have traditionally utilized hexavalent chromium to prevent corrosion and oxidation of steels. Due to the high performance nature of these coatings, work to approve alternate coatings has been difficult. To date, most attempts have failed. The process of developing a novel hexavalent-chromium-free aluminum sacrificial paint will be summarized here, including the history, failures and key factors.

#### Chromate / phosphate coatings

In the 1950s, development of viable flight gas turbine engines was hindered by the lack of coatings to protect steels from salt corrosion and heat. Apart from aluminum-flake filled stoving enamels, there were few coatings that would remain well-bonded to steels at the temperatures being seen in these innovative designs.

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In 1963, Teleflex Inc. (North Wales, PA) filed for a patent for a glassy, heat-curable, aqueous binder that combined phosphoric acid with 5-6 wt% chromic acid (50,000 to 60,000 ppm Cr<sup>+6</sup>).<sup>\*\*\*</sup> By 1965, Pratt, Allison (now Rolls-Royce), and GE were

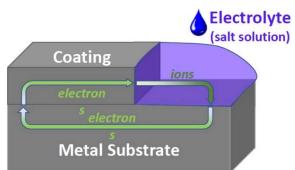
protecting steel shafts and compressor cases in their turbines with a coating of aluminum powder dispersed in this binder. The Al-filled slurry sprayed like a paint, yet when baked at 600°F, it limited oxidation and salt corrosion of steels up to 1000°F.

Soon it was discovered that the aluminum-chromate/phosphate coating would become electrically conductive when it was baked for 90 minutes at 1050°F or lightly blasted with fine grit or beads. When conductive, the coating was galvanically sacrificial (Fig. 1).

In 1970, Teleflex introduced a chromate / phosphate sealer for burnished (sacrificial) SermeTel<sup>™</sup> W creating a sealed coating system, SermeTel<sup>™</sup> 725, with longer life in high salt environments. In 1975, Teleflex's Sermatech Division introduced Process 5375, mechanically polished smooth SermeTel<sup>™</sup> W that increased flight turbine efficiency and reduced fuel consumption by as much as 2%. SermeTel<sup>™</sup> 5380DP, a smoother, sealed and more corrosion resistant coating that did not need polishing, supplanted Process 5375 in 1980. Also that year, Coatings for Industry patented Alseal<sup>™</sup> 500, which cured at temperatures low enough to allow its use on aluminum and HSLA steel.

By the 1980s, chromate/phosphates had become a firmly established standard and were specified by every major turbine engine OEM and many airframe OEMs (Fig. 2). A chromate/phosphate binder was utilized in sealers and other functional coatings, under several trade names.

Aluminum chromate/phosphate coatings (2 mil / 50 µm on steel) met the required specifications. Corrosion resistance per ASTM B117 was met with no rust over 1000 hr of 5% salt fog. Heat / salt corrosion resistance testing produced no rust after 10 cycles. Maximum continuous and peak heat





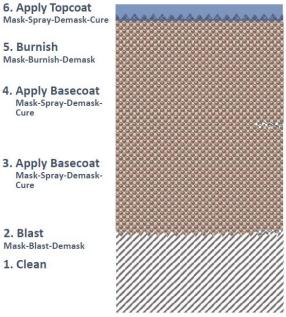


Figure 2 – Schematic of the chromate/phosphate system.

resistance were 1050°F (566°C) and 1150°F (621°C), respectively. Subjecting the coating to either hot lubricating oil or hot deionized water had no effect, and a 90° bend test around a ¼" mandrel produced no flaking.

There was a hidden hazard, however. The heat-curable ceramic binder that Allen had patented combined chromium trioxide (chromic acid) and phosphoric acid. While trivalent chromium ( $Cr^{+3}$ ) was known to be a nutrient essential for human health, chromic acid (hexavalent chromium -  $Cr^{+6}$ ) was known to pose health risks. That is why OSHA limited Permissible Exposure Limits (PEL) for workers to no more than 0.1 mg  $Cr^{+6}/m^3$ , requiring protection and monitoring when there was >0.052 mg  $Cr^{+6}/m^3$ .

#### Elimination of hexavalent chromium

The simple goal was to remove hexavalent chromium from the coating system. Recognizing the risks posed by  $Cr^{+6}$ , around 1990, companies began developing Cr-free alternatives to the Allen binder. In 1993, Solar Turbines was granted the first patent for a Cr-free aluminum-phosphate coating composition. It was stabilized with vanadium pentoxide (V<sub>2</sub>O<sub>5</sub>), a toxin. By 2011,

<sup>\*\*\*</sup> Charlotte Allen, "Inorganic Coating and Bonding Composition," US Patent 3,248,251, April 16, 1966.





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sixteen U.S. patents had been granted for acid-phosphate binder chemistries containing no Cr<sup>+6</sup> that could be used to make a heat-curable aluminum-ceramic. Unfortunately, every Cr-free acid-phosphate had/have limited shelf-life (no more than a few hours) and reacted with bare, blasted steel.

By 2000,  $Cr^{+6}$  in the water supply was indicted as the cause of increasing cancer rates in Hinckley, California. The fight to bring the company responsible for the contamination to justice was dramatized in the film "Erin Brockavitch" and the hazards of  $Cr^{+6}$ were no longer a secret. At the same time, airborne  $Cr^{+6}$  was linked to increased rates of lung cancer. A study published in 2004 estimated that exposure at 0.1 mg  $Cr^{+6}/m^3$  led to an excess risk of lung cancer death that exceeded 1 in 10. In 2005, about one million U.S. workers were exposed to  $Cr^{+6}$ . Thus, it was back to square one.

Blocked by patents on Cr-free acid-phosphate binders, Coatings for Industry (CFI), in 2001, developed a heat-curable aluminumfilled inorganic silicate (water-glass) slurry that could be baked to form an aluminum-ceramic coating. The single component slurry had a one-year shelf-life. It did not react with steels and, when cured, could be made electrically conductive and galvanically sacrificial by the same means now used for Al-chromate/phosphates. In 2003, CFI filed for a patent on its innovative slurry and in 2010, Praxair was granted a patent on its own aluminum-silicate. The patent for the CFI Al-silicate was finally issued in 2014.

At the same time, regulations became stricter. Acknowledging the risks of  $Cr^{+6}$ , in 2006, OSHA lowered the PEL for  $Cr^{+6}$  to 5 µg/m<sup>3</sup>, a 20-fold reduction in the 8-hr limit per employee. Employers had to protect and monitor workers when exposure was a mere 2.5 µg of  $Cr^{+6}$  per cubic meter of air. The European Chemicals Agency (ECHA), classified chromium trioxide as a "substance of very high concern" and proposed to restrict its use in the EU. ECHA's REACh regulations became law in 2017. Users the were required to be authorized (for a substantial fee) to import materials containing >1000 ppm of  $Cr^{+6}$ , registered to use it and report how much they use (Al-chromate/phosphates contain 30K - 55K ppm  $Cr^{+6}$ .)

#### DoD Efforts - Propulsion Environmental Working Group

Anticipating OSHA's lowering of the PEL for Cr<sup>+6</sup>, in 2005, the US Department of Defense (DoD) established the Propulsion Environmental Working Group (PEWG), with the credo, "Industry and government leaders striving to insert clean, safe, and costeffective sustainment technology improvements in the propulsion community." PEWG's goal was to produce "The Ultimate Green Engine" made with parts that lasted its designed lifetime, contained no toxic materials, and built, maintained and reworked without using or releasing any hazardous materials, with valuable components and/or materials recoverable at life's end.

#### Table 1 – Aluminum-ceramic coatings circa 2005.

	<u>Binder</u>	%Cr+6	ppm
<ul> <li>Al-Ceramic Sacrificial Basecoat</li> </ul>			
SermeTel <sup>™</sup> W/962 (Praxair)	chromate/phosphate	3.3 wt. %	30,000
Alseal <sup>™</sup> 518/519 (Coatings For Industry)	chromate/phosphate	3.3 %	30,000
Ceral™ 114* (Morant GmbH, CeralUSA)	chromate/phosphate	(?) ~1.5 %	(?) 15,000
SermeTel CF 1725 (Praxair)	phosphate-borate	0 %	0
Alseal 5000 (now Alseal 5K)	sodium-lithium silicate	0 %	0
Ceramic Topcoat			
SermeTel 570A/TCS-565 (Praxair)	chromate/phosphate	<5.0 wt. %	<50,000
Alseal 598	chromate/phosphate	5.1 %	51,000
VPW 350/Ceral 50*	chromate/phosphate	(?) <0.1 %	(?) 1,000
SermeTel CF 1726 (Praxair)	acid-phosphate	0 %	0
Alseal 5200 (now discontinued)	inorganic silicate	0 %	0
(* According to SDS contains "chromium oxide	(r0 ")		

(\* According to SDS contains "chromium oxide, CrO<sub>3</sub>")

In 2005, there were a number of commercial aluminum-ceramic coating processes available with varying amounts of Cr<sup>+6</sup> in their formulations, as shown in Table 1. Shelf life remained a problem with the alternatives.





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Using specifications for Cr<sup>+6</sup> slurries as a guide, PEWG sought a Cr-free candidate(s) that matched SermeTe<sup>™</sup>I W's performance as a benchmark. Finding no Cr-Free match in all tests, including Cr-free aluminum-phosphate and silicate systems (Table 2), PEWG sup-ported development of a "low Cr<sup>+6</sup>" Al-chromate / phosphate.

Our efforts at CFI continued to focus on a system that was completely Cr<sup>+6</sup>-free. A proprietary system consisting of an aluminum-ceramic sacrificial basecoat using a sodium-lithium silicate binder (Alseal<sup>™</sup> 5000, later referred to as 5K). Two topcoat sealers were considered: (1) Alseal<sup>™</sup> 5200, a sodium-lithium silicate and (2) Alseal<sup>™</sup> 5KTC, using Al<sup>+3</sup>-acid-phosphate chemistry.

The PEWG data in Table 2 shows instability of silicate-based topcoats in fluids. This is further shown in Fig. 3, at left, where the vapor from an open beaker of moderately warm water dissolved the sealer. On the other hand, Fig. 3, at right, shows that an acid-phosphate topcoat stabilized the same Al-silicate under identical test conditions.

JTP Section		SermeTel <sup>®</sup> W	Cr-Free aluminum- phosphate	Cr-Free aluminum- silicate
4.1	Electrical Conductivity	~	1	✓
4.2	Salt Fog Corrosion (Unburnished/No scribe)	~	~	×
4.2	Salt Fog Corrosion (Burnished/Scribe X/with and without topcoat)	$\checkmark$	~	~
4.3	Cyclic Heat/Salt Spray Corrosion Test (Burnished/Scribe X)	$\checkmark$	~	~
4.4	Abrasion Resistance		~	x
4.4	Chipping Resistance	$\checkmark$	✓	✓
4.4	Particle Erosion	$\checkmark$	~	×
4.5	Thermal Stability	✓	✓	✓
4.6	Elevated Temperature Material Compatibility	✓	~	~
4.7	Humidity Exposure	~	x	~
4.7	Pencil hardness after HE	$\checkmark$	×	~
4.8	Fluid Immersion (Hot H2O)	~	~	×
4.8	Fluid Immersion (Hot Lube Oil)	~	1	x
4.8	Fluid Immersion (Anti/deicers)	✓	2	x
4.8	Fluid Immersion (Hydraulic fluid)	~	~	×
4.8	Fluid Immersion (Fuel)	✓	~	x
4.9	Adhesion: 17-4PH	x	×	~
4.9	Adhesion: 4340	x	×	~
4.9	Adhesion: 4130	✓	x	~
4.9	Adhesion: IN718	x	×	~
4.9	Adhesion: A286	x	×	~
4.10	Strip-Ability		1	
4.11	Hydrogen Embrittlement	1	1	1
4.12	Electrochemical	✓	~	✓
4.13	Constant Amplitude Axial Fatigue (High Cycle Fatigue)	✓	~	~
4.14	Stress Corrosion Cracking	$\checkmark$	~	~

Table 2 – Pass/fail characteristics of PEWG analysis of aluminum-ceramic coating systems.





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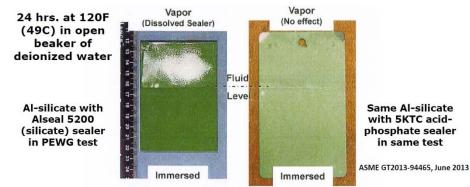


Figure 3 – Performance of Cr-free sealers in deionized water at 120°F (49°C).

Unfortunately, a Cr-free coating that had passed lab tests failed in a power generation turbine operating with wet compression. The failure led the OEM to develop a new laboratory test to screen the Cr-free systems. It involved a 100-hr partial immersion in deionized water at 176°F (80°C) in sealed container. The result for the system in question, a Cr-free Al-silicate basecoat, burnished and sealed with a Cr-free acid-phosphate topcoat, is shown in Fig. 4.

Our laboratory looked at alternate approaches of using these new Al-acid phosphate sealers, keeping in mind the PEWG support of low  $Cr^{+6}$  chemistries with  $Cr^{+6}$  levels well below the REACh limits of less than 1000 ppm. The system schematically shown in Fig. 5 was



Figure 4 – Exfoliation failure of Cr-free Al-silicate basecoat, sealed with a Cr-free acid-phosphate topcoat in hot water immersion test.

the result. The Alseal  $^{\text{TM}}$  5K Cr<sup>+6</sup>-free aluminum-silicate basecoat was followed by immersion in a <u>heat-cured</u> acid-phosphate solution, which stabilized the basecoat in hot water or water vapor testing. Two chemistries were tested: (1) a completely Cr-free solution, Alseal  $^{\text{TM}}$  5K-S1 or (2) a very low Cr<sup>+6</sup> (< 1 ppm) trivalent chromium acid-phosphate solution, Alseal  $^{\text{TM}}$  5K-T3. This step was followed by an abrasive burnish, which activated the cathodic response of the aluminum-silicate. Finally, a second application of the smooth acid-phosphate topcoat, the very low Cr<sup>+6</sup> Alseal  $^{\text{TM}}$  5K-T3 containing trivalent chromium completed the process. Figure 6 shows the results after 1000 hr of hot immersion testing of this system in 176°F (80°C) deionized hot water, a ten-fold increase in test duration. No coating failure was observed.

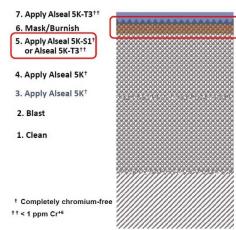


Figure 5 – Schematic of the REACh compliant systems.

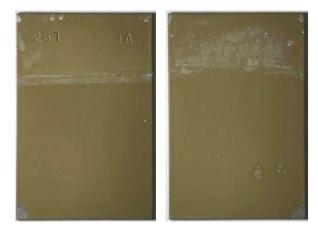


Figure 6 – Results of 1000 hr of hot water immersion test for Alseal<sup>™</sup> REACh-compliant system.





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#### Summary

Virtually Cr-free slurries have been developed, consisting of a completely Cr-free aluminum-ceramic sacrificial basecoat, based on sodium-lithium silicate chemistry (Alseal<sup>™</sup> 5K) plus ceramic sealers, consisting of (a) a completely Cr-free aluminum-acid phosphate chemistry (Alseal<sup>™</sup> 5K-S1) or (b) an aluminum-trivalent chromium acid phosphate chemistry (Alseal<sup>™</sup> 5K-T3) containing less than 1 ppm Cr<sup>+6</sup> well below the REACh compliance-limit of 1000 ppm. It appears that these nearly Cr-Free materials can be combined into multi-layer coating systems that are a safe alternative to aluminum sacrificial paints (ASPs) that contain hazardous chromium trioxide (hexavalent chromium) which are currently widely used in aerospace.

#### About the author



Bruce McMordie is Director of Alseal Operations at Coatings for Industry, Inc. (cficoatings.com) in Souderton, Pennsylvania. He has over 40 years' experience developing and implementing coating systems to protect components in turbine and industrial equipment. He spent more than 30 years with Sermatech International (now Praxair Surface Technologies) and 10 years now with Coatings For Industry. Bruce holds 18 coating related patents.